

Start instructions

Hot cathode tubes are thin-walled, highly evacuated glass tubes. Treat them carefully as there is a risk of implosion.

- Do not subject the tube to mechanical stresses.
- Do not subject the connection leads to any tension.
- The tube may only be used with tube holder D. If voltage or current is too high or the cathode is at the wrong temperature, it can lead to the tube becoming destroyed.
- Do not exceed the stated operating • parameters.
- Only change circuit with power supply equipment switched off.
- Only exchange tubes with power supply equipment switched off.
- When the tube is in operation, the stock of the tube may get hot.
- If necessary, allow the tube to cool before dismantling.

Technical data

Filament voltage: ≤ 7.0 V AC/DC

Anode voltage: 0–5000 V DC

Anode current: typ. 0.15 mA at 4000 V DC

Lattice constant of graphite:

d10= 0.213 nm

d11= 0.123 nm

Distance from graphite target to fluorescent screen: 125 ± 2 mm.

Fluorescent screen: 100 mm dia.

Glass bulb:130 mm dia.

Total length:260 mm dia.

Purpose of the set-up

The electron diffraction tube illustrates the wave nature of electrons by allowing observation of interference caused by a beam of electrons passing through a polycrystalline graphite target on a fluorescent screen (Debye-Scherrer diffraction). The wavelength of the electrons can be calculated for various anode voltages from the radius of the diffracted rings and the distance between the crystal layers in the graphite. The tube also confirms the de Broglie hypothesis. The electron diffraction tube is a highly evacuated tube with an electron gun consisting of a pure tungsten heater filament and a cylindrical anode all contained in a clear glass bulb. The electrons emitted by the heated cathode are constrained to a narrow beam by an aperture and are then focused by means of an electron-optical system. The resulting tight, monochromatic beam then passes through a micro-mesh nickel grating situated at the aperture of the gun. Onto this grid, a thin layer of polycrystalline graphitised carbon has been deposited by vaporisation. This layer affects the electrons in the beam much like a diffraction grating. The result of this diffraction is seen in the form of an image comprising two concentric rings that become visible on the fluorescent screen. A spot resulting from the undeflected electron beam continues to be visible at the center of the rings.

A magnet is also supplied with the tube. This allows the direction of the electron beam to be changed, which may be necessary if the graphite target has slight damage as a result of the manufacturing process or due to later overheating.

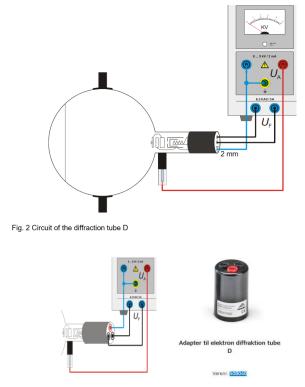
Functionality / experimenting

The graphite foil on the diffraction grating is only a few layers of molecules thick and any current greater 0.2 mA can cause its destruction. The internal resistor is there to prevent damage to the graphite foil. The graphite target itself should be monitored throughout the experiment. If the graphite target starts to glow, the anode must immediately be disconnected from its power supply If the diffraction rings are not satisfactorily visible, the electron beam can be redirected by a magnet so that it passes through an undamaged region of the target.

Setting up the experiment as in Fig. 2.

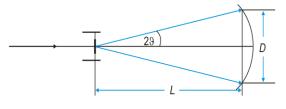
Connect the negative pole of the anode supply via the 2-mm socket.

Apply the heater voltage and wait about 1 minute for the heater temperature to achieve thermal stability, and then apply an anode voltage of 4 kV. Determine the diameter D of the diffraction rings.



(alternative connection with adapter for tube D)

Two diffraction rings appear on the fluorescent screen centered on the undeflected beam in the middle. The two rings correspond to Bragg reflections from atoms in the layers of the graphite crystal lattice. Changing the anode voltage causes the rings to change in diameter. Reducing the voltage makes the rings wider. This supports de Broglie's postulate that the wavelength increases as momentum is reduced.



- a) Bragg equation: $\lambda = \sin 2d\vartheta$
- λ = wavelength of the electrons
- $\vartheta\text{=}$ glancing angle of the diffraction ring
- d= lattice plane spacing in graphite
- L = distance between sample and screen
- D = diameter of the diffraction ring
- R= radius of the diffraction ring:

Tan2
$$\vartheta$$
 = D/2xL, $\cdot \lambda = d \cdot \frac{R}{L}$

b) De-Broglie equation:

$$\lambda = \frac{h}{p}$$

H = Planck's constant

p= momentum of the electrons:

$$e \cdot U = \frac{p^2}{2 \cdot m}, \ \lambda = \frac{h}{\sqrt{2 \cdot m \cdot e \cdot U}}$$

m = electron mas,

e= electron charge